# 31. Diastereoselective Synthesis of Cyclododeca-1,6-diallenes ( = Cyclododeca-1,2,6,7-tetraenes) 

by Christoph Boss and Reinhart Keese*<br>Departement für Chemie und Biochemie der Universität Bern, Freiestrasse 3, CH-3012 Bern

(23.VIII.96)


#### Abstract

The synthesis of substituted cyclododeca-1,6-diallenes ( $=$ cyclododeca-1,2,6,7-tetraenes) from cyclododeca-5,11-diyne-1,4-diols is described (Schemes 1 and 3). The ca.1:1 mixtures of the stereoisomers of the cyclododeca1,6 -diallenes were formed in high yields from the $c a .1: 1$ diastereoisomer mixtures of the 1,4 -disubstituted cyclododeca- 5,11 -diynes by reactions with $\mathrm{Me}_{2} \mathrm{CuLi}$ or $t-\mathrm{BuMgCl} / \mathrm{Cu}^{1} \mathrm{I}$. In mechanistically relevant experiments with the pure diastereoisomers of 1,4-dimethylcyclododeca-5,11-diyne-1,4-diol, it is demonstrated that the configuration is conserved in these reactions. The first synthesis of a 1 -substituted cyclododeca- 2,8 -diyne bearing only one propargylic leaving group gives access to a mixed 12-membered allen-yne (Scheme 5).


Introduction. - The intriguing selectivity of allenes in their reactions with electrophiles, radicals, and nucleophiles makes them attractive precursors for $\mathrm{C}-\mathrm{C}$ bond forming reactions [1]. They are readily available by a variety of methods, many of which make use of $S_{\mathrm{N}} 2^{\prime}$-type reactions of propargylic precursors or the MeLi -induced opening of geminal dibromocyclopropanes [2-16]. The chiral nature of disubstituted allenes, predicted by van't Hoff, led to the investigation of their configuration which clearly shows that 1,3 -disubstituted allenes exist as enantiomers. For cyclic compounds, the minimal ring size, which would allow the strainless incorporation of the linear $\mathrm{C}=\mathrm{C}=\mathrm{C}$ structure and isolation of these allenes was of particular interest: cyclonona-1,2-diene is the smallest stable cyclic allene which has been isolated [1] [17].

For cyclic diallenes, it has been established by Sondheimer that the cyclododeca-1,7diallene ( $=$ cyclododeca-1,2,7,8-tetraene) $\mathbf{1}$ exist in both, the meso and the racemic $C_{2}$ form [18]. In the cyclodecane series, only the meso form of the cyclodeca-1,6-diallene ( = cyclododeca-1,2,6,7-tetraene) $\mathbf{2}$ had been prepared and its structure determined [19].

rac-1

meso-1

meso-2

Recently, we developed efficient synthetic routes to cyclododeca-2,8-diynes with substituents at the propargylic positions of the two alkyne groups [20] and subsequently explored their transformation into cyclic 1,6 -diallenes. Due to their ready accessibility
and their stability, the cyclododeca-5,11-diyne-1,4-diols 3a and 7a with tertiary- and secondary-alcohol groups, respectively, were chosen as precursors for the synthesis of the cyclododeca-1,6-diallenes 4 and 8 .

Results and Discussion. - When a ca. 1:1 mixture of the diastereoisomers of 1,4-dimethylcyclododeca-5,11-diyne-1,4-diyl diacetate (3b) was treated with $\mathrm{Me}_{2} \mathrm{CuLi}$, the cyclododeca-1,6-diallene 4 was formed as a ca.1:1 diastereoisomer mixture in $90 \%$ yield (Scheme 1). The ${ }^{13} \mathrm{C}$-NMR of 4 showed 16 lines, and the GC analysis revealed two signals in a ca. 1:1 ratio which, according to GC/MS, showed both the same fragmentation pattern from $m / z 216$ for $M^{+1}$ ).

Scheme 1

a) $\mathrm{Ac}_{2} \mathrm{O}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{DMAP}$, r.t., 10 h. b) $\mathrm{LiCuMe}_{2}, \mathrm{Et}_{2} \mathrm{O},-15^{\circ}$.

When the same reaction conditions were applied to the diastereoisomer mixture (ca. $1: 1$ ) of the acyclic 3,6-dimethylocta-1,7-diyne-3,6-diyl diacetate ( $\mathbf{5 b}$ ), the diallene $\mathbf{6}$ was also obtained as a ca. 1:1 diastereoisomer mixture in $91 \%$ yield (Scheme 2). This was apparent from the double set of signals in the ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum of 6 . Due to its thermal lability, $\mathbf{6}$ could not be analyzed by GC [23] [24].

Scheme 2


meso-6

rac-6
a) $\mathrm{Ac}_{2} \mathrm{O}, \mathrm{Py}, \mathrm{DMAP}, 0^{\circ}, 1.5$ h, then r.t., 24 h. b) $\mathrm{LiCuMe}_{2}, \mathrm{Et}_{2} \mathrm{O},-15^{\circ}$.

In a further transformation, a ca. 1:1 diastereoisomer mixture 7b reacted with $t$ - BuMgCl and $\mathrm{Cu}^{\mathrm{I} I}$ to the $1,8-\mathrm{di}($ tert-butyl)cyclododeca-1,6-diallene $\mathbf{8}$ in $72 \%$ yield [8] (Scheme 3). The diastereoisomeric meso- and rac-8 were formed in a ca. 1:1 ratio, which

[^0]
## Scheme 3



a) $\square$ 7a $\begin{aligned} & \mathrm{R}=\mathrm{H} \\ & \square \text { b } \mathrm{R}=\mathrm{Ms}\end{aligned}$

rac-8

$$
\text { a) } \mathrm{Et}_{3} \mathrm{~N}, \mathrm{MsCl}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{THF}, 0^{\circ} \text {. b) } t \mathrm{BuMgCl}, \mathrm{Cul}, \mathrm{THF},-30^{\circ} \text {. }
$$

was confirmed by the double set of signals in the ${ }^{13} \mathrm{C}-\mathrm{NMR}$ and the GC analysis with a chiral stationary phase.

In a mechanistically relevant experiment, separate transformations of the pure diastereoisomers of $\mathbf{3 b}$, obtained by acetylation of the pure diastereoisomers of $\mathbf{3 a}$, led to products each containing a single diastereoisomer of 4 . The ${ }^{13} \mathrm{C}$-NMR spectra showed different sets of 8 signals. According to GC analyses on a chiral stationary phase, the precursor meso-3b was transformed into meso-4, whereas rac-4 was obtained from the rac-precursor ${ }^{2}$ ).

These results might be interpreted as follows: the $S_{\mathrm{N}} 2^{\prime}$ reactions can overall follow a trans or a cis pathway, with the trans pathway usually being favored [7] [10] [12] (trans refers to the incoming nucleophile and the leaving group). For a trans pathway, it is reasonable to assume that these substitution reactions occur in a plane perpendicular to the idealized plane of the ring system, because only cis substitutions could occur in-plane. For the formation of meso- 4 by two sequential trans- $S_{\mathrm{N}} 2^{\prime}$ substitutions, the pathway for the second propargylic displacement can be pictured as shown in $9 \mathbf{a}$. In the case of rac-3b, the second trans substitution reaction with $\mathrm{Me}_{2} \mathrm{CuLi}$ may proceed via an intermediate with a conformation as depicted in $\mathbf{9 b}$. Inspection of models indicate that there is severe interaction between the Me group introduced in the first $S_{\mathrm{N}} 2^{\prime}$ reaction and the $\mathrm{CH}_{2}$ group in the $\gamma$-position. The molecule might, therefore, adopt the less strained conformation 9 c where the $\mathrm{CH}_{2}$ groups now interfere with the trans-mode of $\mathrm{Me}^{-}$attack in the propargylic position. Alternatively, the stereochemical pathway of the transformation $\mathbf{3 b}$ to $\mathbf{4}$ could proceed similarly to that proposed for 7 b to $\mathbf{8}$ (see below). The stereospecific results imply that in rac- $\mathbf{3 b}$, the expected steric interaction is not strong enough to change the mode of the $S_{\mathrm{N}^{\prime}} 2^{\prime}$ reaction from trans to cis as described for other examples [7]. These facts only show that the configuration is preserved in the reactions of meso- and rac-3b but do not answer the question of whether the chirality transfer occurs via a cis or a trans mode in these cyclic systems [25] ${ }^{3}$ ).

[^1]

9a


9b


9c

Applying these concepts of steric hindrance to the reaction of $\mathbf{7 b}$ to the diallene $\mathbf{8}$, the stereochemical course of the substitution can be rationalized by assuming that a conformational change will occur after the first $S_{\mathrm{N}}{ }^{2}$ reaction. This would lead to an intermediate with a quasi-equatorial $t$-Bu group and an essentially unhindered alkyne for the second trans-substitution proceeding via 10 b to 10 c and to the conformation of meso-8 depicted in Scheme 4. This interpretation is supported by AM1 calculations of a tert-butylcyclododeca-1-allen-7-yne, which gave a conformation with a quasi-equatorial tert-butyl group. The transformation of $\operatorname{rac}-7 \mathrm{~b}$ to $\mathrm{rac}-\mathbf{8}$ can similarly be analyzed according to this scheme.

Scheme 4. Conformational Analysis of the Transformation of meso-7b to meso-8



As a first example of the 'missing link' between the dialkynes $\mathbf{3 a}, \mathbf{b}$ and $\mathbf{7 a}, \mathbf{b}$ and the diallenes 4 and 8 , the cyclic allen-yne compound $\mathbf{1 2}$ was prepared from the mono-functionalized dialkyne 11 b [20].

Concluding Remarks. - The transfer of chirality from the propargylic positions in the cyclic alkynes by $S_{\mathrm{N}}{ }^{2}$-type reactions to the allenes can be explained in terms of the conformational constraints of the ring system and by steric interactions in the course of the second $S_{\mathrm{N}} 2^{\prime}$ reaction. In the case of $\mathbf{3 b}$, it has been proven that the diastereoisomers

Scheme 5

a) $\mathrm{MsCl}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{THF}, 0^{\circ}, 2.5$ h. b) ${ }^{\mathrm{t}} \mathrm{BuMgCl}, \mathrm{Cul}, \mathrm{THF},-30^{\circ}$.
are transformed into the diastereoisomers of the tetramethylcyclododeca-1,6-diallene 4 stereospecifically with conservation of the diastereoisomeric relationship.

The ready access of the cyclic 1,6 -dialienes $\mathbf{4}$ and $\mathbf{8}$ and the cyclic allen-yne $\mathbf{1 2}$ calls for a study of their transannular reactions.

We are grateful to the Stipendienfonds der Basler Chemischen Industrie for a stipend to C.B. and to the Swiss National Science Foundation for generous support of this work (project No. 20.37270.93 and 20-43565.95). The authors would like to thank $P$. Hübscher and D. Lehmann who have contributed with their excellent experimental skills to the success of these investigations and $A$. Saxer for thorough GC analyses.

## Experimental Part

General. Chemicals were purchased from commercial suppliers and used without further purification. MeLi (Fluka pract.) was used as a 1.6 m soln. in $\mathrm{Et}_{2} \mathrm{O}$. THF was dried by distillation from $\mathrm{Na}, \mathrm{Et}_{2} \mathrm{O}$ from NaH . DMAP $=4$-(dimethylamino) pyridine. After workup by pouring the mixture onto sat. $\mathrm{NH}_{4} \mathrm{Cl}$ soln. and extraction with $\mathrm{Et}_{2} \mathrm{O} /$ pentane $1: 1$, the solns. were dried $\left(\mathrm{MgSO}_{4}\right)$. TLC: silica-gel plates $S I L$ G/UV $V_{254}$ (Macherey \& Nagel). GC: Hewlett-Packard-HP-5890 instrument (He, 43 kPa ) with a HP-5-Ultra capillary column (length 10 m , i.d. $0.2 \mathrm{~mm})$ and a temp. program $40-220^{\circ}\left(3^{\circ} / \mathrm{min}\right)$; chiral analyses at 20 kPa with a modified cyclodextrin as chiral stationary phase; column $A(10 \mathrm{~m}$, d.f. $0.25 \mu \mathrm{~m}$, i.d. 0.3 mm$) 100 \%$ heptakis $\{2,3$-di- $O$-acetoxy- 6 - $O$-[(tert-butyl)dimethylsilyll $\}-\beta$-cyclodextrin in OV 1701 and variable temp. programs; $t_{\mathrm{R}}$ in min. Prep. HPLC: 715004 ET--250/ 10-Nuc. 50-7 column (Macherey-Nagel); flow $12 \mathrm{ml} / \mathrm{min}$. M.p.: Büchi-510 melting-point apparatus; uncorrected. IR Spectra: in $\mathrm{CHCl}_{3}$; Perkin-Elmer-782-IR spectrophotometer. NMR Spectra: in $\mathrm{CDCl}_{3}$ or $\left(\mathrm{D}_{6}\right)$ acetone; Bruker-AC-300 spectrometer ( ${ }^{1} \mathrm{H}, 300 \mathrm{MHz} ;{ }^{13} \mathrm{C}, 75 \mathrm{MHz}$ ); chemical shifts $\delta$ in ppm rel. to internal $\mathrm{CHCl}_{3}(\delta$ 7.27) or ( $\mathrm{D}_{6}$ ) acetone ( $\delta 2.04$ ) for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and $\mathrm{CDCl}_{3}(\delta 77.0)$ or ( $\mathrm{D}_{6}$ )acetone ( $\delta 29.8$ ) for ${ }^{13} \mathrm{C}-\mathrm{NMR}$; stack = heavily overlapping signals. MS: Varian-MAT-CH7A (70 eV, EI) and Fisons-Autospec-Q spectrometer; in $m / z\left(\right.$ rel. $\%$ ) GC/MS: VG-Autospec spectrometer. Reactions were normally performed under Ar or $\mathrm{N}_{2}$.

1. Allene Precursors 3b, 5b and 11b. 1,4-Dimethyl-cyclododeca-5,11-diyne-1,4-diyl Diacetate (3b). Under the same conditions as described for $\mathbf{5 b}$ (see below), 0.2 g ( 0.91 mmol ) of 1,4-dimethylcyclododeca-5,11-diyne-1,4-diol (3a) [20] gave $0.083 \mathrm{~g}(30 \%)$ of $\mathbf{3 b}$ as a colorless liquid. Better yields were obtained by dissolving $\mathbf{3 a}(0.143 \mathrm{~g}$, $0.65 \mathrm{mmol})$ in $\mathrm{Ac}_{2} \mathrm{O}(4 \mathrm{ml})$ followed by slow addition of $\mathrm{Et}_{3} \mathrm{~N}(0.262 \mathrm{~g}, 2.29 \mathrm{mmol})$ and DMAP $(0.079 \mathrm{~g}$, 0.65 mmol ) and stirring at r.t. for 10 h . Workup and chromatography with $\mathrm{AcOEt} /$ hexane $2: 1$ yielded 0.168 g $(85 \%)$ of 3b as a colorless liquid. The pure diastereoisomers of 3b were obtained by separate transformation of the pure diastereoisomers of 3a, obtained by crystallization [20]. $R_{f}$ (meso/rac) 0.60 (AcOEt/hexane 2:1). GC (meso/rac): $t_{R} 33.98,34.17$ (ca. 1:1). IR (meso/rac): 2850s, 2225w, $1710 \mathrm{~m}, 1330 \mathrm{~s}, 1240 \mathrm{vs}, 1170 \mathrm{~s}, 1100 \mathrm{~s}, 1080 \mathrm{vs}$, $1020 s, 930 s$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right)\right.$ acetone; meso/rac): $1.56-1.78(2 s+$ stack; 10 H$) ; 1.81-1.89(d, 2 \mathrm{H}) ; 2.020(s, 3 \mathrm{H})$, $2.023(s, 3 \mathrm{H}) ; 2.12-2.32$ (stack, 4 H$) ; 2.32-2.43$ (stack, 2 H ). ${ }^{13} \mathrm{C}$-NMR ( $\left(\mathrm{D}_{6}\right)$ acetone): rac-3b: $19.10(t) ; 21.75(q)$; $26.50(q) ; 27.67(t) ; 38.82(t) ; 76.58(s) ; 82.15(s) ; 86.20(s) ; 167.43(s) ;$ meso-3b $19.00(t) ; 21.40(q) ; 25.93(q)$; $27.56(t) ; 38.53(t) ; 76.59(s) ; 82.04(s) ; 86.33(s) ; 167.41(s)$. MS (meso/rac): no $M^{+}, 234(5), 220(3), 219(7), 203(58)$, 192(29), 187(30), 173(11), 169(11), 159(22), 145(16), 143(18), 131(26), 119(13), 117(13), 105(19), 91(24), 77(13), 43(100).

3,6-Dimethylocta-1,7-diyne-3,6-diyl Diacetate (5b). To a soln. of 3,6-dimethylocta-1,7-diyne-3,6-diol ( 0.5 g , 3.01 mmol ; 5a) [20] in pyridine ( 25 ml ) was slowly added $\mathrm{Ac}_{2} \mathrm{O}(0.97 \mathrm{~g}, 9.06 \mathrm{mmol}$ ) and DMAP ( 0.15 g , 1.23 mmol ) at $0^{\circ}$ and stirred for 1.5 h . The mixture was warmed to r.t. and stirring continued for 24 h . Workup and chromatography with $\mathrm{Et}_{2} \mathrm{O}$ gave $0.45 \mathrm{~g}(60 \%)$ of $\mathbf{5 b}$. White solid. M.p. $52^{\circ} . R_{\mathrm{f}} 0.68\left(\mathrm{Et}_{2} \mathrm{O}\right) . \mathrm{GC}: t_{\mathrm{R}} 12.45$, 12.54 (52:48). IR: 3310s, 3000 m , 2940 m , 2120w, 1740 vs , $1445 \mathrm{~m}, 1370 \mathrm{~s}$, $1240 \mathrm{vs}, 1175 \mathrm{~m}, 1060 \mathrm{~m}, 1015 \mathrm{~m}, 940 \mathrm{~m}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 1.59(s, 6 \mathrm{H}) ; 1.92(s, 6 \mathrm{H}) ; 1.92-2.11$ (stack, 4 H$) ; 2.49(s, 2 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right):$ major isomer: $21.80(q) ; 26.42(q) ; 36.05(t) ; 73.66(s) ; 74.10(d) ; 83.32(s) ; 169.20(s) ;$ minor isomer: $21.80(q) ; 26.32(q)$; $35.99(t) ; 73.64(s) ; 74.06(s) ; 83.29(s) ; 169.20(s)$. GC-MS: $208\left([M-43]^{+}, 0.5\right), 166(24), 165\left([M-43-43]^{+}, 27\right)$, 149(10), 138(15), 137(10), 133(17), 117(11), 116(11), 105(13), 91(18), 80(19), 79(33), 69(9), 53(13), 43(100). Anal. calc. for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{4}$ (250.29): C67.18, H7.25; found: C67.46, H7.43.

Cyclododeca-2,8-diyn-1-yl Methanesulfonate (11b). As described in [20], cyclododeca-2,8-diyn-1-ol ( 0.1 g , $0.57 \mathrm{mmol} ; 11 \mathrm{a})$ [20] was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(7 \mathrm{ml}) \mathrm{at}-5^{\circ}$. After the addition of $\mathrm{Et}_{3} \mathrm{~N}(0.234 \mathrm{~g} 2.31 \mathrm{mmol})$ and $\mathrm{MsCl}(0.216 \mathrm{~g}, 1.88 \mathrm{mmol})$, the mixture was stirred for 4 h at $0^{\circ}$. Workup with 2 N HCl and extraction with $\mathrm{Et}_{2} \mathrm{O}$ gave $0.132 \mathrm{~g}(91 \%)$ of 11 b as a white powder which could be used without further purification. $R_{f} 0.73\left(\mathrm{Et}_{2} \mathrm{O}\right)$. IR: $2930 \mathrm{vs}, 2860 \mathrm{~s}, 2235 \mathrm{~m}, 1740 \mathrm{~m}, 1680 \mathrm{~m}, 1315 \mathrm{~m}, 1165 \mathrm{vs}, 1090 \mathrm{~m}, 1010 \mathrm{~m}, 965 \mathrm{~s}, 900 \mathrm{vs} .{ }^{1} \mathrm{H}-\mathrm{NMR}$ ( $\mathrm{D}_{6}$ )acetone): 1.59-1.99 (stack, 6H); 2.00-2.40 (stack, 8 H ); $3.19(s, 3 \mathrm{H}) ; 5.24(d m, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ ((D. $\mathrm{D}_{6}$ )acetone): $19.10(t)$; $19.30(t) ; 19.39(t) ; 24.59(t) ; 27.33(t) ; 28.40(t) ; 35.42(t) ; 39.13(q) ; 73.13(d) ; 78.08(s) ; 81.48(s) ; 82.32(s) ; 90.95(s)$. MS: $255\left([M+1]^{+}, 0.5\right), 254\left(M^{+}, 4\right), 221(5), 220(23), 206(15), 205(77), 189(5), 175(20), 158(24), 157(32), 147(22)$, 143(33), 131(40), 130(52), 129(77), 128(50), 117(55), 115(66), 105(41), 103(33), 91(87), 81(27), 79(100), 78(36), $77(60), 67(39), 65(42), 55(40), 51(36), 41(61), 39(66)$.
2. Acylic and Cyclic Diallenes 4, 5, and $\mathbf{8}$ and Allen-yne 12. 1,3,6,8-Tetramethylcyclododeca-1,2,6,7-tetraene (4). To a slurry of $\mathrm{CuI}(0.244 \mathrm{~g}, 1.28 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(2.6 \mathrm{ml})$ was slowly added $\mathrm{MeLi}(1.60 \mathrm{ml}, 1.6 \mathrm{~m}, 2.56 \mathrm{mmol})$ at $-15^{\circ}$. After 20 min , meso- $\mathbf{3 b}(0.098 \mathrm{mg}, 0.324 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(0.8 \mathrm{ml})$ was added and stirred at $-10^{\circ}$ for 30 min . The mixture was warmed to r.t. and stirred for another 24 h . Workup and chromatography with pentane gave meso-4 ( $63 \mathrm{mg}, 90 \%$ ) as a colorless liquid. Under the same conditions, the transformation of rac-3b gave rac-4 in $91 \%$ yield.
meso-4: $R_{\mathrm{r}} 0.61$ (pentane). GC (cf. Footnote 2): $t_{\mathrm{R}} 26.49$; column $A$ (isothermal at $120^{\circ}$ ): $t_{\mathrm{R}} 47.70$. IR: 2960 vs , $2940 \mathrm{vs}, 2900 \mathrm{vs}, 2860 \mathrm{vs}, 1965 \mathrm{w}, 1470 \mathrm{~s}, 1455 \mathrm{~s}, 1370 \mathrm{~s} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 1.63(\mathrm{~s}, 6 \mathrm{H}) ; 1.67(\mathrm{~s}, 6 \mathrm{H}) ; 1.68-1.80$ (stack, 4 H ); 1.93-2.06 (stack, 4 H ); 2.14-2.36 (stack, 4 H$) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 19.48(q) ; 19.70(q) ; 25.61(t)$; $31.82(t) ; 32.25(t) ; 96.35(s) ; 97.64(s) ; 199.40(s) . \mathrm{MS}: 217\left([M+1]^{+}, 17\right), 216\left(M^{+}, 100\right), 201(27), 187(13), 173(20)$, 161(6), 160(8), 159(49), 146(9), 145(40), 133(17), 131(15), 119(33), 105(18), 91(16), 77(12), 67(5), 41(15). HR-MS: $216.1876\left(\mathrm{C}_{16} \mathrm{H}_{24}^{+}\right.$; calc. 216.1878).
rac-4: $R_{\mathrm{f}} 0.61$ (pentane). GC (cf. Footnote 2): $t_{\mathrm{R}} 25.82$; column $A\left(60-200^{\circ}\left(2^{\circ} / \mathrm{min}\right)\right): t_{\mathrm{R}} 47.91,48.50(1: 1)$. IR: $2960 \mathrm{vs}, 2940 \mathrm{vs}, 2900 \mathrm{vs}, 2860 \mathrm{vs}, 1965 w, 1470 \mathrm{~s}, 1455 \mathrm{~s}, 1370 \mathrm{~s} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 1.628(s, 6 \mathrm{H}) ; 1.634(\mathrm{~s}, 6 \mathrm{H})$; $1.73-2.03$ (stack, 8 H$) ; 2.22(d, 2 \mathrm{H}) ; 2.25(d, 2 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 18.28(q) ; 20.97(q) ; 26.37(t) ; 32.03(t) ;$ $33.29(t) ; 96.67(s) ; 97.18(s) ; 199.65(s)$. MS: $217\left([M+1]^{+}, 18\right), 216\left(M^{+}, 100\right), 201(24), 187(10), 173(17), 161(5)$, 160(7), 159(43), 146(7), 145(35), 133(15), 131(13), 119(30), 105(16), 91(14), 77(10), 67(5), 55(8), 41(15). HR-MS: $216.1877\left(\mathrm{C}_{16} \mathrm{H}_{24}^{+}\right.$; calc. 216.1878).

4,7-Dimethyldeca-2,3,7,8-tetraene (6). As described for 4, with $\mathbf{5 b}(1.44 \mathrm{~g}, 5.77 \mathrm{mmol})$. Workup and chromatography with pentane gave $0.85 \mathrm{~g}(91 \%)$ of 6 . Colorless liquid. $R_{\mathrm{f}} 0.72$ (pentane). GC: 6 was not stable under GC conditions [23] [24]. IR: $2950 \mathrm{vs}, 2920 \mathrm{vs}, 2900 \mathrm{vs}$, $2870 \mathrm{vs}, 1970 \mathrm{~m}, 1470 \mathrm{~s}, 1445 \mathrm{~s}, 1370 \mathrm{~s}, 1270 \mathrm{~m}, 1145 \mathrm{~m}, 995 \mathrm{~m}$. ${ }^{1} \mathrm{H}$-NMR $\left(\mathrm{CDCl}_{3}\right): 1.64(d d, 6 \mathrm{H}) ; 1.69(d, 6 \mathrm{H}) ; 2.04(s, 4 \mathrm{H}) ; 5.01($ stack, 2 H$) .{ }^{13} \mathrm{C}$-NMR $\left(\mathrm{CDCl}_{3}\right)$ : major isomer: $14.83(q) ; 19.24(q) ; 32.02(t) ; 84.93(d) ; 98.44(s) ; 201.93(s) ;$ minor isomer: $14.87(q) ; 19.28(q) ; 32.08(t) ; 84.96(d)$; $98.50(s) ; 201.93(s) . \operatorname{MS}: 161\left([M-1]^{+}, 1\right), 151(6), 148(8), 147\left([M-15]^{+}, 93\right), 137(12), 133(62), 119(17), 105(29)$, 91(24), 85(17), 71(34), 57(76), 43(100).

1,8-Di(tert-butyl) cyclododeca-1,2,6,7-tetraene (8). To 7b [20] ( $84 \mathrm{mg}, 0.24 \mathrm{mmol}$ ) and Cul ( $93 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) in THF ( 5 ml ) was added at $-30^{\circ}$ a freshly prepared $t-\mathrm{BuMgCl}$ soln. in $\mathrm{Et}_{2} \mathrm{O}(1 \mathrm{ml}, 1.0 \mathrm{mmol}$; obtained from Mg ( $292 \mathrm{mg}, 12 \mathrm{mmol}$ ) and $t-\mathrm{BuCl}\left(926 \mathrm{mg}, 10 \mathrm{mmol}\right.$ ) in $\mathrm{Et}_{2} \mathrm{O}(10 \mathrm{ml})$ ). The mixture was stirred for 30 min , then warmed to r.t. and stirred for 1 h . Workup and chromatography with pentane gave $47.2 \mathrm{mg}(72 \%)$ of 8 as a colorless liquid which formed white crystals at $-18^{\circ}$. M.p. $48^{\circ} . R_{\mathrm{f}} 0.55$ (pentane). GC: $t_{\mathrm{R}} 37.69,37.58$ (1:3); column $A$ (isothermal at $110^{\circ}$ ): $t_{\mathrm{R}} 59.93$ ( $74 \%$; meso $+1 / 2 \mathrm{rac}$ ), $63.24(26 \% ; 1 / 2 \mathrm{rac}):$ IR: $2975 \mathrm{vs}, 2960 \mathrm{vs}, 2920 \mathrm{vs}$, $2880 \mathrm{vs}, 2850 \mathrm{vs}, 1955 \mathrm{~m}, 1720 \mathrm{w}, 1465 \mathrm{vs}, 1450 \mathrm{vs}, 1360 \mathrm{~s}, 1340 \mathrm{~m}, 1290 \mathrm{~m}, 1240 \mathrm{~s}, 1205 \mathrm{~s}, 1090 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right):$ $1.03(s, 18 \mathrm{H}) ; 1.78-2.34$ (stack, 12 H ); 4.94-5.16 (stack, 2 H ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : major isomer: $26.58(t)$; $27.07(t) ; 29.33(q) ; 29.73(t) ; 33.29(s) ; 92.65(d) ; 113.10(s) ; 199.84(s)$; minor isomer: $26.58(t) ; 27.17(t) ; 29.46(q)$; $29.73(t) ; 33.39(s) ; 92.91(d) ; 113.30(s) ; 199.87(s)$. MS: $273\left([M+1]^{+}, 5\right), 272\left(M^{+}, 22\right), 257(5), 216(22), 215(46)$, $201(20), 193(12), 173(24), 160(20), 159(67), 131(29), 105(17), 95(17), 91(19), 85(36), 57(100), 43(33)$. HR-MS: $272.2504\left(\mathrm{C}_{20} \mathrm{H}_{32}^{+}\right.$; calc. 272.2504).

1-(tert-Butyl) cyclododeca-1,2-dien-7-yne (12). As described for 7b, with 11 b ( $0.13 \mathrm{~g}, 0.512 \mathrm{mmol}$ ): 0.06 g ( $54 \%$ ) of 12. Colorless liquid. $R_{\mathrm{f}} 0.39$ (pentane). GC: $t_{\mathrm{R}} 29.86$. IR: $2940 \mathrm{~m}, 2900 \mathrm{~m}, 2820 \mathrm{~s}, 2320 \mathrm{~m}, 1950 \mathrm{~m}, 1450 \mathrm{~s}$, $1435 \mathrm{vs}, 1390 \mathrm{~m}, 1360 \mathrm{vs}, 1330 \mathrm{~m}, 1110 \mathrm{vs}, 1020 \mathrm{~s}, 980 \mathrm{~m}, 900 \mathrm{~s} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 1.02(\mathrm{~s}, 9 \mathrm{H}) ; 1.48-1.98$ (stack, $8 \mathrm{H}) ; 2.00-2.43$ (stack, 6 H ); $5.10($ stack, 1 H$) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) ; 17.89(t) ; 19.19(t) ; 25.15(t) ; 25.18(t) ; 26.44(t)$ : $27.92(t) ; 28.54(t) ; 29.29(q) ; 33.53(s) ; 81.22(s) ; 83.08(s) ; 94.56(d) ; 112.74(s) ; 199.41(s)$. MS: $216\left(M^{+}, 9\right)$, 201(45), 173(26), $159(100), 145(25), 131(63), 117(49), 105(27), 91(49), 79(25), 77(23), 67(20), 57(32), 41(48)$. HR-MS: $216.1875\left(\mathrm{C}_{16} \mathrm{H}_{24}^{+}\right.$; calc. 216.1878).

## REFERENCES

[1] W. R. Moore, R. C. Bertelson, J. Org. Chem. 1962, 27, 4182.
[2] P. Crabbé, E. Barreiro, J.-M. Dollat, J.-L. Luche, J. Chem. Soc., Chem. Commun. 1976, 183.
[3] P. Rona, P. Crabbé, J. Am. Chem. Soc. 1968, 90, 4733.
[4] J.-L. Luche, E. Barreiro, J.-M. Dollat, P. Crabbé, Tetrahedron Lett. 1975, 4615.
[5] P. Crabbé, H. Carpio, J. Chem. Soc., Chem. Commun. 1972, 904.
[6] P. Vermeer, I. Meijer, L. Brandsmaa, Recl. Trav. Chim. Pays-Bas 1975, 94, 112.
[7] P. Vermeer, H. Westmijze, H. Kleijn, L. A. van Dijck, Recl. Trav. Chim. Pays-Bas 1978, 97, 56.
[8] J.-L. Moreau, M. Gaudemar, J. Organomet. Chem. 1976, 108, 159.
[9] K. A. Parker, J. J. Petraitis, Tetrahedron Lett. 1977, 4561.
[10] I. Marek, P. Mangeney, A. Alexakis, J. F. Normant, Tetrahedron Lett. 1986, 27, 5499.
[11] A. Alexakis, A. Commercon, J. Villiéras, J. F. Normant, Tetrahedron Lett. 1976, 2313.
[12] A. Alexakis, P. Mangeney, J. F. Normant, Tetrahedron Lett. 1985, 6, 4197.
[13] D. J. Pasto, R. H. Shults, J. A. McGrath, A. Waterhouse, J. Org. Chem. 1978, 43, 1382.
[14] A. Claesson, L.-I. Olsson, J. Am. Chem. Soc. 1979, 101, 7302.
[15] L. A. van Dijck, B. J. Lankwerden, J. G. C. M. Vermeer, A. J. M. Weber, Recl. Trav. Chim. Pays-Bas 1971, $90,801$.
[16] L. Skattebol, Acta Chem. Scand. 1963, 17, 1683.
[17] R. P. Johnson, Mol. Struct. Energ. 1986, 3, 85.
[18] P. J. Garratt, K. C. Nicolaou, F. Sondheimer, J. Am. Chem. Soc. 1973, 95, 4582.
[19] H. Irngartinger, H.-U. Jäger, Tetrahedron Lett. 1976, 3595.
[20] C. Boss, R. Keese, Helv. Chim. Acta 1996, 79, 2164.
[21] T. Tabuchi, J. Inanaga, M. Yamaguchi, Tetrahedron Lett. 1986, 27, 5237.
[22] T. Tabuchi, J. Inanaga, M. Yamaguchi, Chem. Lett. 1987, 2275.
[23] D. J. Pasto, Tetrahedron 1984, 40, 2805.
[24] S. Braverman, Y. Duar, Tetrahedron Lett. 1978, 1493.
[25] O. Eisenstein, G. Procter, J. D. Dunitz, Helr. Chim. Acta 1978, 61, 2538.


[^0]:    ${ }^{1}$ ) The reaction of $\mathbf{3 b}$ with $\mathrm{H}^{-}$as the nucleophile using $\mathrm{Sml}_{2} / \mathrm{Pd}^{0}$ remained unsuccessful [21] [22].

[^1]:    ${ }^{2}$ ) The GC of meso- and rac-4 contained each an additional peak ( $t_{\mathrm{R}} 20.5 \mathrm{~min}$ ) of $30-40 \%$ intensity, whereas no by-product could be observed in their ${ }^{13} \mathrm{C}$-NMR spectra. GC/MS Spectra of this peak showed $m / z 216$ for $M^{+}$, identical to that of 4 . Therefore, the by-product must correspond to an isomer of 4 of unknown structure [23]. This indicates a limited thermal stability of the diastereoisomers of 4 . No additional peaks could be detected in the GC of $\mathbf{8}$ and $\mathbf{1 2}$, where the allenic unit is substituted by a $t$-Bu group.
    ${ }^{3}$ ) It is questionable, whether the model calculations which show an angle of $120^{\circ}$ for the addition of $\mathrm{H}^{-}$to $\mathrm{HC} \equiv \mathrm{CH}[25]$ are stereoelectronically relevant for the transformation of propargylic compounds like $\mathbf{3 b}$ with $\mathrm{R}_{2} \mathrm{CuLi}$.

